

Synthesis of 6,8,9-Tri- and 2,6,8,9-Tetrasubstituted Purines by a Combination of the Suzuki Cross-coupling, N-Arylation, and Direct C-H Arylation Reactions

Igor Čerňa, Radek Pohl, Blanka Klepetářová, and Michal Hocek*

Institute of Organic Chemistry and Biochemistry, Academy of Sciences of the Czech Republic, Gilead Sciences & IOCB Research Center, CZ-16610, Prague 6, Czech Republic

hocek@uochb.cas.cz

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Novel practical methodology of synthesis of a several types of di-, tri-, and tetraarylpurine derivatives by a combination of regioselective Suzuki cross-coupling reactions and/or Cu-catalyzed N-arylation with direct C–H arylations was developed. 6,8-Diaryl- and 2,6,8-triaryl-9-isopropylpurines were prepared by one or two cross-couplings of 6-chloro- or 2,6-dichloro-9-isopropylpurine with arylboronic acids followed by Pd-catalyzed C–H arylation by aryl halides to position 8. 6-Chloropurine and adenine underwent Cu-catalyzed N-arylation to position 9 with boronic acids, followed by cross-coupling with AlMe₃ and/ or C–H arylation to obtain 8,9-diaryl-6-methylpurines or 8,9-diaryladenines (accompanied by products of partial N-arylation of adenine in position 6). The methodology is suitable for construction of small libraries of modified purines.

Introduction

2,6,9- or 6,8,9-Trisubstituted and 2,6,8,9-tetrasubstituted purines display a wide range of biological activities,¹ such as inhibition of protein kinases² or tubulin polymerization,³ and antagonist effects to receptors,⁴ etc. Reversine is a purine derivative causing a dedifferentiation of muscle cells to progenitor cells⁵ and inducing polyploidization⁶ of cancer cells via inhibition of Aurora kinases. 6-Arylpurines are of particular importance due to anti-HCV, cytostatic, and antimycobacterial

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activities.⁷ Large combinatorial libraries of several types of triand tetrasubstituted purines have been prepared by heterocyclizations⁸ or by regioselective nucleophilic substitutions of dihalopurines with amines (sometimes in combination with one cross-coupling reaction).^{9,10} In contrast to simple, efficient, and

^{*} To whom correspondence should be addressed. Phone: +420 220183324; fax: +420 220183559.

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Synthesis of Tri- and Tetrasubstituted Purines

regioselective introduction of N- or C-substituents,¹⁰ regioselective introduction of C-substituents by cross-coupling reactions¹¹ is more problematic. Although 2,6- and 6,8-dihalopurines give regioselective cross-couplings with most types of organometallics and were used for the synthesis of series of several types of disubstituted purine bases and nucleosides,12 reactions of 2,6,8-trichloropurine proceed unselectively, giving mixtures of products.¹³ Therefore, we have started out developing the methodology of direct C-H arylation^{14,15} in purines as a new strategy complementary to the cross-couplings applicable for regioselective multiple substitutions of purine. Recently, we have reported¹⁶ on the new C-H arylations of 6,9-disubstituted purines to position 8 by diverse aryl halides under Pd catalysis in absence of ligands (in analogy to Bellina's protocol¹⁵). As a

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proof of principle, this method was applied¹⁵ in combination with two regioselective cross-coupling reactions to get two examples of 2,6,8-trisubstituted purines. Later on, we¹⁷ and others¹⁸ have further developed the C-H arylation for modification of unprotected nucleosides (i.e., adenosine).

Several types of cytostatic natural products (e.g., colchicine, podophyllotoxin, and combretastatine) displaying antimitotic effect through inhibition of tubulin polymerization possess a common structural feature of two heavily methoxylated aromatic rings in the vicinity. Many diverse non-natural compounds of such and related structures (diarylpyrroles,¹⁹ -triazoles,²⁰ -imidazoles,²¹ -thiophenes²² etc.) were prepared and were also found to be potent inhibitors of tubulin polymerization. Myoseverin³ is a related cytostatic purine derivative, 2,6-bis[(4-methoxybenzyl)amino]-9-isopropylpurine, with the same mechanism of action. Our own study²³ on analogous 2,6-diaryl- and -dibenzylpurines did not show significant activities except for 2,6diarylethynylpurines. Now, we have decided to further combine the stuctural features of three classes of cytostatic compounds-2,6,9-trisubstituted purines, 6-arylpurines, and combretastatine analogues-and use a purine ring as a scaffold for attachment of two or three methoxylated phenyl groups in other positions and their combinations. Therefore, our target compounds were novel 9-isopropyl-6,8-diarylpurines, -2,6,8-triarylpurines, 6-methyl-8,9-diarylpurines, and 8,9-diaryladenines, and our synthetic methodology of choice involved a combination of the cross-coupling reactions in positions 2 and/or 6, Narylations in position 9, and C-H arylation in position 8.

Results and Discussion

As mentioned above, our synthetic strategy toward the target compounds was based on a combination of the Pd-catalyzed Suzuki-Miyaura cross-coupling reactions of (di)halopurines with arylboronic acids, Pd-calatyzed C-H arylations of purines with aryl halides, and Cu-catalyzed N-arylations of 9H-purines with arylboronic acids. Therefore a set of five methoxylated phenylboronic acids (1a-e) and four methoxylated phenyl iodides and bromides (2a-d) has been chosen (Chart 1) as the reagents for the particular arylations in different position.

The first class of compounds of our interest were 9-isopropyl-6,8-diarylpurines envisaged to be prepared by a two-step sequence of the Suzuki reaction in position 6 followed by the C-H arylation in position 8. Thus 6-chloro-9-isopropylpurine (3) underwent the Pd-catalyzed Suzuki-Miyaura cross-coupling reactions with four arylboronic acids 1a-d under classical conditions 12c,23 in toluene in presence of Pd(PPh₃)₄ and K₂CO₃ to get a series of 6-aryl-9-isopropylpurines (4a-d, Scheme 1) isolated in good yields (except for the 3,4,5-trimethoxyphenyl derivative 4d isolated in rather moderate yield of 66%). In the second step, each of these intermediates 4a-d was used in a series of C-H arylation¹⁶ experiments with methoxyphenyl halides (2a-d) to generate a small library of 16 6,8-diarylpu-

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SCHEME 1. Preparation of 6,8-disubstituted-9-isopropyl-9*H*-purines



 $\begin{array}{l} (i) \ R^{1}\text{-}B(OH)_{2} \ (\textbf{1a-1d}, \ 1.3 \ equiv), \ Pd(PPh_{3})_{4} \ (0.024 \ equiv), \\ K_{2}CO_{3} \ (1.25 \ equiv), \ Toluene, \ 19 \ h, \ 100^{\circ}C. \\ (ii) \ R^{2}\text{-}X \ (\textbf{2a-2d}, \ 2 \ equiv), \ Pd(OAc)_{2} \ (0.05 \ equiv), \ Cs_{2}CO_{3} \\ (2.5 \ equiv), \ Cul \ (3 \ equiv), \ DMF, \ 60 \ h, \ 160^{\circ}C. \end{array}$

rines (**5xy**, Scheme 1). The direct C–H arylation reactions were performed under the previously reported¹⁶ optimized conditions in the presence of Pd(OAc)₂ (without any ligand) and excess of CuI and Cs₂CO₃ in DMF at 160 °C for 60 h. The yields of

the second step varied between 23 and 80% depending on the aryl halide used. Somewhat surprisingly we did not observe significant differences in reactivity between substituted phenyl iodides and bromides, but the least reactive was the most electron-rich 3,4,5-trimethoxyphenyl iodide (yields 23-40%). The moderate yields were mostly caused by incomplete conversion of the starting materials rather than by formation of sideproducts.¹⁶ Formation of minor 8,8'-bis(purine) side-products¹⁶ has been observed in several cases with yields of <5%, but in the case of reaction of 4a or 4b with 2a, the bis(purines) were isolated in 24 and 10%, respectively (see Supporting Information), in addition to the desired major products 5aa (64%) and **5ba** (80%). The overall yields of the desired title 6.8-diarylpurines 5xy over the two steps ranged from moderate (15%) to excellent (71%), and the methodology was quite efficient for a facile generation of the 4×4 library of derivatives.

The second class of compounds under study were 9-isopropyl-2,6,8-triarylpurines. To access these compounds, we decided to combine two regioselective Suzuki-Miyaura cross-couplings with direct C-H arylation. In our previous communication, we have demonstrated, as a proof-of-principle, on two examples that, starting from 9-substituted 2,6-dichloropurine, one can perform consecutively two cross-couplings followed (without isolation of intermediates) by the C-H arylation. We have now applied the approach for the synthesis of our target 2,6,8,9tetrasubstituted purines. Thus, the starting 2,6-dichloro-9isopropylpurine 6 was reacted successively in one pot with two arylboronic acids, and then, without isolation, the reaction mixture was just filtered through a pad of celite, evaporated, and directly used for C-H arylation reaction (Scheme 2). Both reactions were performed under the same conditions as discussed above in the previous series of compounds. The first aryl group was regioselectively introduced to position 6 (using arylboronic acids 1a, 1b, and 1e), the second to position 2 (using 1a and **1b**), and the third one to position 8 (using aryl halides **2a** and 2b). Using this methodology, we prepared 9 examples of desired 2,6,8,9-tetrasubstituted purines (7xyz) in very good total yields (average yield 48% over three steps). Reactions proceeded relatively cleanly with only one major product, easily separable from minor impurities (presumably mono- and diarylated purines) by column chromatography.

In the third series of compounds, 6-methyl-8,9-diarylpurines, we used a combination of N^9 -arylation, cross-coupling with trimethylaluminum, and direct C-H arylation. This time the intermediates after each step were isolated from the reaction mixture on a pad of silica gel due to incompatible reagents used in each step. The N^9 -arylations of 6-chloropurine 8 with arylboronic acids (1a, 1b) were performed according to the Gundersen procedure²⁴ in presence of anhydrous $Cu(OAc)_2$, phenanthroline and molecular sieves in dichloromethane gave desired products 9a and 9b in ca. 50% yields (Scheme 3). Methylation of intermediates 9a and 9b with trimethylaluminum was performed in analogy to literature²⁵ in the presence of Pd(PPh₃)₄ in THF to give the 6-methyl-9-arylpurines 10a and 10b in 63 and 80% yields, respectively. The last step, C-H arylation in position 8 of 10a,b with aryl halides 2a, 2b, and 2d proceeded smoothly under the same conditions as above to give six examples of the target compounds 11x, y in 52-97%

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SCHEME 2. Consecutive synthesis of 2,6,8-trisubstituted-9-isopropyl-9*H*-purines



(i) 1. $R^{1}B(OH)_{2}$ (**1a,1b, 1e,** 1.05 equiv), Pd(PPh₃)₄ (0.05 equiv), K₂CO₃ (3 equiv), Toluene, 100°C, 18h. 2. $R^{2}B(OH)_{2}$ (**1a,1b,** 2 equiv) 100°C, 12h. (ii) R^{2} -X (**2a,2b**, 2 equiv), Pd(OAc)₂ (0.05 equiv), Cs₂CO₃ (2.5 equiv), Cul (3 equiv), DMF, 60 h, 160°C.

yields of the third step. Again, reactions with 5-iodo-1,2,3trimethoxybenzene **2d** gave lower conversions to give the products **11ad** and **11bd** in 53 and 52%, respectively.

In the last series we focused on synthesis of 8,9-disubstitutedadenines. The first step was N^9 -arylation of adenine with arylboronic acids **1a** and **1b** according to the literature²⁶ in presence of Cu(OAc₂)•H₂O and TMEDA (Scheme 4). These reactions proceeded very slowly to give the 9-aryladenines **12a,b** in moderate yields of 37 and 28%, respectively (Table 2), after separation from unreacted starting material by a simple filtration through a short pad of silica gel. Considering the low cost of the starting adenine and simple separation of products, the low-



two byproducts were also isolated arising from single (14aa) and double (15aa) Cu-mediated N⁶-arylation of 13aa (Scheme 4). This phenomenon was also observed in direct C-H arylation on adenosine,¹⁷ where partial (in this case only one-fold) N^6 arylation of the product also took place. Under the standard conditions mentioned above, we isolated 45% of desired product 13aa and 29 (14aa) and 17% (15aa) of the byproducts (Table 1, entry 1), which were easily separable from each other by flash chromatography. Several optimization experiments have been performed in order to suppress the side reaction. Shortened reaction time led to lower conversion (entry 2), whereas a lower temperature (120 °C, Table 1, entry 3) slightly increase the yield of desired product 13aa. Decreasing or increasing of the amount of copper iodide did not bring any improvements (Table 1, entries 4 and 5). The use of piperidine as base (entry 6) gave the best selectivity and yield of formation of 8-aryladenine 13aa (79%), accompanied only by 10% of minor product 14aa (15aa was not observed). We took advantage of the good separation of the three products (13, 14, and 15) to diversity-oriented synthesis of several derivatives in parallel and used the conditions giving good yields of all of them (Table 1, entry 3) in arylation of **12a,b** with a series of aryl halides **2a**, **2b**, and 2d. These six arylation reactions gave 18 compounds (Scheme 4, Table 2). The yields of the 8,9-diaryladenines 13x,y ranged from 21 to 50%, whereas the N-arylated products 14 and 15 were isolated in 5-31% yields. It is interesting to mention that although the arylation of 12 with mono- and dimethoxyphenyl halides 2a and 2b produced the 8-aryladenines 13 as major products, the reaction with trimethoxyphenyl iodide 2d led to preferential formation of N^6 ,8-diarylated products 14ad and 14bd (Table 2, entries 3 and 6).

All the intermediates and final compounds were fully characterized by NMR spectroscopy (including C,H-HSQC and C,H-HMBC experiments for complete assignment of all signals,

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SCHEME 4. Synthesis of 8,9-disubstituted-adenines

(i) R¹-B(OH)₂ (1a,1b, 2 equiv), Cu(OAc)₂.H₂O (1.0 equiv), TMEDA (2 equiv), CH₃OH:H₂O=4:1, air atm. 24h, r.t. (ii) R²-X (2a,2b,2d, 2 equiv), Pd(OAc)₂ (0.05 equiv),Cul (3 equiv), Cs₂CO₃ (2.5 equiv), DMF, 120°C, 42 h.

TABLE 1	Reaction (of 12a	with 2a	Ontimization	of	conditions
IADLE I.	Reaction of	01 12a	with $2a$.	Opumization	UI.	continuous

Entry	<i>T</i> (°C)	<i>t</i> (h)	base	CuI (equiv)	13aa (%)	14aa (%)	15aa (%)
1	160	48	Cs ₂ CO ₃	3	45	29	17
2	160	21	Cs ₂ CO ₃	3	39	27	15
3	120	42	Cs ₂ CO ₃	3	46	24	16
4	120	42	Cs_2CO_3	1	37	28	24
5	120	42	Cs_2CO_3	6	21	17	18
6	150	14	piperidine	3	79	10	0

see Supporting Information). Selected examples of compounds were also characterized by single-crystal X-ray diffraction. Figure 1 shows the ORTEP plots for structures of compounds **7eaa**, **13ad**, and **15aa**. It demonstrates a low steric hindrance in the 2,6,8-arylpurines, where in **7eaa** all the three aryl groups are nearly coplanar to the purine ring. On the other hand, in 8,9-diarylpurines, the two bulky adjacent aryl groups cannot adopt planar conformation, and the one in position 9 is perpendicular to the purine plane (**13ad** and **15aa**), similarly to the two aryl groups of diarylamino group in compound **15aa**.

All the title substituted purines 4, 5, 7, 10, 11, 13, 14, and 15 were subjected to biological activity screening. The cytostatic activity in vitro (inhibition of cell growth)^{7a} was studied on the following cell cultures: (i) mouse leukemia L1210 cells (ATCC CCL 219); (ii) human promyelocytic leukemia HL60 cells (ATCC CCL 240); (iii) human cervix carcinoma HeLaS3 cells (ATCC CCL 2.2), and (iv) human T lymphoblastoid CCRF-CEM cell line (ATCC CCL 119). The cytostatic activities of compounds were determined by XTT analysis. None of the studied compounds showed any significant cytostatic effect at $c = 1-10 \,\mu$ M (depending on the solubility in water). In addition, inhibition of polymerization of tubulin²⁷ was performed on all the compounds and again, no significant effect was found. The very limited solubility in water was definitely a limitation for this class of compounds.

In conclusion, we have developed the synthetic strategy for the preparation of several novel classes of tri- and tetrasubstituted purines by a combination of the Suzuki–Miyaura crosscoupling reactions of (di)chloropurines with arylboronic acids in position(s) 6 (and 2), the Cu-catalyzed N-arylation in position 9 and direct C–H arylation in position 8. The 2–3 step synthetic sequences in principle can be performed without isolation of intermediates (but not one-pot due to incompatibility of the reagents and solvents in each step). The approach is efficient and suitable for parallel synthesis of combinatorial libraries of purines bearing several aryl groups in different position. The compounds under study were inactive in cytostatic activity screening and in tubulin assay; however, the cross-couplings, N and C–H arylations should be suitable for combination with amination(s) in position(s) 6 (and/or 2) to generate the second generation of compounds with a better solubility in water.

Experimental Section

General Procedure for Suzuki Cross-coupling Reactions of 3 with 1a-d. Toluene (5 mL) was added through a septum to an argon-purged vial containing 6-chloro-9-isopropyl-9*H*-purine (3, 196.6 mg, 1 mmol), Pd(PPh₃)₄ (27.7 mg, 0.024 mmol), arylboronic acid (1a-d, 1.3 mmol), and K₂CO₃ (172.7 mg, 1.25 mmol). The reaction mixture was heated to 100 °C for 19 h. The solvent was evaporated under reduced pressure. Products were isolated by flash column chromatography (gradient elution hexanes \rightarrow ethyl acetate/ hexanes 1:1).

9-Isopropyl-6-(4-methoxyphenyl)-9*H***-purine (4a).** Yield 98%, white crystals from CHCl₃/heptane, 89–92 °C. ¹H NMR (600 MHz, CDCl₃): 1.68 (d, 6H, $J_{vic} = 6.8$, $(CH_3)_2$ CH); 3.91 (s, 3H, CH₃O); 4.98 (h, 1H, $J_{vic} = 6.8$, $CH(CH_3)_2$); 7.08 (m, 2H, H-3',5'); 8.16 (s, 1H, H-8); 8.81 (m, 2H, H-2',6'); 8.97 (s, 1H, H-2). ¹³C NMR (151 MHz, CDCl₃): 22.6 ((*C*H₃)₂CH); 47.1 (*C*H(CH₃)₂); 55.4 (CH₃O); 114.0 (CH-3',5'); 128.4 (C-1'); 130.9 (C-5); 131.4 (CH-2',6'); 141.4 (CH-8); 151.9 (C-4); 152.0 (CH-2); 154.4 (C-6); 161.9 (C-4'). MS (FAB), *m/z* (% relative intensity): 184 (10), 227 (40), 269 (MH⁺, 100). HR MS (MH⁺) 269.140558 (calcd for C₁₅H₁₆N₄O 269.140236). Anal. calcd for C₁₅H₁₆N₄O: C, 67.15; H, 6.01; N, 20.88. Found: C, 67.07; H, 5.90; N, 20.69.

General Procedure for C-H Arylation of 6-Aryl-9-isopropylpurines 4a-d with 2a-d. DMF (6 mL) was added through a septum to an argon-purged vial containing a 6-aryl-9-isopropyl-

Entry	\mathbf{R}^1	12 <i>x</i>	R ²	Product (Yield)			
		(Yield)		13 <i>xy</i>	14xy	15xy	
1	1 2 3 MeO	12a (37%)	4-MeO-Ph	13aa (46 %)	14aa (24 %)	15aa (16 %)	
2			3,4-(MeO) ₂ -Ph	13ab (26 %)	14ab (12 %)	15ab (18 %)	
3			3,4,5-(MeO) ₃ -Ph	13ad (21 %)	14ad (24 %)	15ad (15 %)	
4	- Vy		4-MeO-Ph	13ba (50 %)	14ba (27 %)	15ba (18 %)	
5	5 MeO OMe	12b (28%)	3,4-(MeO) ₂ -Ph	13bb (26 %)	14bb (5 %)	15bb (9 %)	
6			3,4,5-(MeO) ₃ -Ph	13bd (23 %)	14bd (31 %)	15bd (10 %)	

TABLE 2. Synthesis of 8,9-disubstituted-adenines

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FIGURE 1. ORTEP drawings of crystal structures of 7eaa (a), 13ad (b), and 15aa (c) with atom numbering scheme. Thermal ellipsoids are drawn at the 50% probability level.

9*H*-purine (1 mmol), Pd(OAc)₂ (11.2 mg, 0.05 mmol, 5 mol %), CuI (571.3 mg, 3 mmol), aryl halide (**2a**–**d**, 2 mmol), and Cs₂CO₃ (814.6 mg, 2.5 mmol). Reaction mixture was stirred at 160 °C for 60 h and, after cooling to rt, it was diluted with chloroform (20 mL) and solvents were evaporated under reduced pressure. Products were isolated by flash column chromatography (gradient elution hexanes \rightarrow ethyl acetate/hexanes 1:1).

9-Isopropyl-6,8-bis(4-methoxyphenyl)-9H-purine (5aa). Yield 52%, white crystals from CHCl₃/heptane, mp 151–152 °C. ¹H NMR (600 MHz, CDCl₃): 1.78 (d, 6H, $J_{vic} = 6.9$, $(CH_3)_2$ CH); 3.89 (s, 3H, CH₃O-4'); 3.91 (s, 3H, CH₃O-4''); 4.83 (h, 1H, $J_{vic} = 6.9$, $CH(CH_3)_2$); 7.06 (m, 2H, H-3',5'); 7.09 (m, 2H, H-3'',5''); 7.68 (m, 2H, H-2'',6''); 8.88 (m, 2H, H-2',6'); 8.93 (s, 1H, H-2). ¹³C NMR (151 MHz, CDCl₃): 21.1 ((CH₃)_2CH); 49.7 (CH(CH₃)_2); 55.3 (CH₃O-4'); 55.5 (CH₃O-4''); 113.9 (CH-3',5'); 114.3 (CH-3'',5''); 122.4 (C-1''); 128.8 (C-1'); 130.9 (C-5); 131.0 (CH-2'',6''); 131.4 (CH-2',6'); 151.0 (CH-2); 153.3 (C-6); 153.9 (C-4); 154.5 (C-8); 161.3 (C-4''); 161.6 (C-4'). MS (FAB), *m/z* (% relative intensity): 134 (35), 290 (5), 303 (6), 319 (5), 333 (59), 345 (6), 359 (5), 375 (MH⁺, 100). HR MS (MH⁺) 375.183958 (calcd for C₂₂H₂₂N₄O₂ 375.182101). Anal. calcd for C₂₂H₂₂N₄O₂: C, 70.57; H, 5.92; N, 14.96. Found: C, 70.59; H, 5.84; N, 14.64.

General Procedure for Consecutive Synthesis of 2,6,8,9-Tetrasubstituted Purines 7xyz. Toluene (3 mL) was added through a septum to an argon-purged vial containing a 2,6-dichloro-9isopropyl-9H-purine (6, 231.1 mg, 1 mmol), arylboronic acid (1a,1b,1e, 1.05 mmol), Pd(PPh₃)₄ (57.8 mg, 0.05 mmol, 5 mol %), and K₂CO₃ (414.6 mg, 3 mmol). Reaction mixture was heated to 100 °C for 18 h. After cooling to rt, the second arylboronic acid (1a,1b, 2 mmol) and toluene (6 mL) were added. The reaction mixture was heated to 100 °C for 12 h, and after being cooled to rt it was diluted with CHCl₃ (100 mL) and filtered through a pad of celite. The solvents were evaporated, and the residue was dried at room temperature under vacuum for 14 h. The crude reaction mixture was transferred into a vial and Pd(OAc)₂ (11.2 mg, 0.05 mmol, 5 mol %), CuI (571.3 mg, 3 mmol), aryl halide (2a,2b, 2 mmol), and Cs₂CO₃ (814.6 mg, 2.5 mmol) were added. The vial was closed with a pressure cap, evacuated and filled with argon. Then DMF (6 mL) was added under argon atmosphere, and the reaction mixture was heated to 160 °C for 60 h. The crude mixture was diluted with CHCl₃ (50 mL), and the solvents were evaporated

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under reduced pressure. Products **7xyz** were isolated by flash column chromatography (gradient elution 5% ethyl acetate in hexanes \rightarrow 30% ethyl acetate in hexanes).

8-(3,4-Dimethoxyphenyl)-9-isopropyl-2,6-bis(4-methoxyphenyl)-9H-purine (7aab). Yield 54%, white crystals from CHCl₃/ heptane, mp 198-201 °C. ¹H NMR (500 MHz, CDCl₃): 1.85 (d, 6H, $J_{\text{vic}} = 6.8$, (CH₃)₂CH); 3.90 (s, 3H, CH₃O-4"); 3.91 (s, 3H, CH₃O-4'); 3.98 (s, 3H, CH₃O-4"''); 3.99 (s, 3H, CH₃O-3"'); 4.86 (h, 1H, $J_{\text{vic}} = 6.8$, $CH(CH_3)_2$); 7.04 (d, 1H, $J_{5''',6'''} = 8.2$, H-5'''); 7.05 (m, 2H, H-3',5'); 7.08 (m, 2H, H-3",5"); 7.27 (dd, 1H, J_{6",5"} $= 8.2, J_{6''',2'''} = 2.0, H-6'''); 7.31 (d, 1H, J_{2''',6'''} = 2.0, H-2'''); 8.64$ (m, 2H, H-2',6'); 9.01 (m, 2H, H-2",6"). ¹³C NMR (125.7 MHz, CDCl₃): 21.3 ((CH₃)₂CH); 49.6 (CH(CH₃)₂); 55.35 and 55.38 (CH₃O-4',4"); 56.09 and 56.13 (CH₃O-3"',4"'); 111.1 (CH-5"'); 112.7 (CH-2"); 113.7 (CH-3',5'); 113.8 (CH-3",5"); 122.3 (CH-6"'); 122.9 (C-1"'); 129.3 (C-5); 129.4 (C-1"); 129.7 (CH-2',6'); 131.5 (CH-2",6"); 131.8 (C-1'); 149.3 (C-3""); 150.8 (C-4""); 152.8 (C-6); 154.3 (C-8); 154.9 (C-4); 157.0 (C-2); 161.1 (C-4'); 161.5 (C-4"'). IR (KBr): 2937, 2909, 2839, 1608, 1566, 1513, 1462, 1376, 1305, 1248, 1165, 1027, 867, 619. MS (ESI), m/z (% relative intensity): 316.4 (3), 469.3 (2), 511.2 (MH⁺, 100). HR MS (MH⁺) 511.23368 (calcd for C₃₀H₃₁N₄O₄ 511.23453). Anal. calcd for C₃₀H₃₀N₄O₄: C, 70.57; H, 5.92; N, 10.97. Found: C, 70.37; H, 6.09; N. 10.62

General Procedure for C-H Arylation of 9-Aryl-adenines 12a,b with 2a, 2b, and 2d. DMF (6 mL) was added through a septum to an argon-purged vial containing a 9-aryl-adenine (1 mmol), Pd(OAc)₂ (11.2 mg, 0.05 mmol, 5 mol %), CuI (571.3 mg, 3 mmol), aryl halide (2 mmol), and Cs₂CO₃ (814.6 mg, 2.5 mmol), and the reaction mixture was heated to 120 °C for 42 h. The crude mixture was diluted with CHCl₃ (50 mL), and the solvents were evaporated under reduced pressure. Products 13xy, 14xy, and 15xy were isolated by flash chromatography on silica gel (gradient elution 40% ethyl acetate in hexanes \rightarrow ethyl acetate).

8,9-Bis(4-methoxyphenyl)-adenine (13aa). Yield 46%, white crystals from CHCl₃/heptane, $R_f 0.17$ (EtOAc), mp 127–130 °C. ¹H NMR (600 MHz, CDCl₃): 3.80 (s, 3H, CH₃O-4'); 3.86 (s, 3H, CH₃O-4''); 6.09 (bs, 2H, NH₂); 6.83 (m, 2H, H-3',5'); 7.00 (m, 2H, H-3'',5''); 7.22 (m, 2H, H-2'',6''); 7.46 (m, 2H, H-2',6'); 8.33 (s, 1H, H-2). ¹³C NMR (151 MHz, CDCl₃): 55.3 (CH₃O-4'); 55.6 (CH₃O-4''); 113.9 (CH-3',5'); 115.0 (CH-3'',5''); 119.0 (C-5); 121.4 (C-1'); 127.6 (C-1''); 128.5 (CH-2'',6''); 130.6 (CH-2',6'); 150.5 (C-8); 152.3 (C-4); 152.9 (CH-2); 154.8 (C-6); 159.7 (C-4''); 160.8 (C-4'). IR (KBr): 3490, 3288, 3151, 2976, 2910, 2841, 1639, 1598, 1567, 1515, 1469, 1449, 1416, 1331, 1304, 1254, 1173, 1102, 1031, 839, 812, 684. MS (ESI), *m/z* (% relative intensity): 288.3 (20), 316.4 (9), 348.2 (MH⁺, 100). HR MS (MH⁺) 348.14454 (calcd for C₁₉H₁₈N₅O₂ 348.14550). Anal. calcd for C₁₉H₁₇N₅O₂•1CHCl₃: C, 51.47; H, 3.89; N, 15.0. Found: C, 51.13; H, 3.83; N, 14.86.

 N^{6} ,8,9-Tris(4-methoxyphenyl)-adenine (14aa). Yield 24%, brownish crystals from CHCl₃/heptane, R_f 0.79 (EtOAc), mp 94–98 °C. ¹H NMR (600 MHz, CDCl₃): 3.82 (s, 3H, CH₃O-4"); 3.83 (s, 3H, CH₃O-4'); 3.87 (s, 3H, CH₃O-4"''); 6.86 (m, 2H, H-3',5'); 6.96 (m, 2H, H-3',5'); 7.02 (m, 2H, H-3''',5'''); 7.27 (m, 2H, H-2''',6'''); 7.51 (m, 2H, H-2",6"); 7.70 (m, 2H, H-2',6'); 7.75 (bs, 1H, NH); 8.48 (s, 1H, H-2). ¹³C NMR (151 MHz, CDCl₃): 55.3 (CH₃O-4"); 55.5 and 55.5 (CH₃O-4',4"'); 114.0 (CH-3",5"); 114.3 (CH-3',5'); 115.0 (CH-3"',5"'); 119.7 (C-5); 121.6 (C-1"); 122.4 (CH-2',6'); 127.8 (C-1""); 128.6 (CH-2"",6""); 130.6 (CH-2",6"); 131.7 (C-1'); 150.3 (C-8); 151.96 and 151.97 (C-4,6); 153.0 (CH-2); 156.0 (C-4'); 159.7 (C-4"'); 160.8 (C-4"). IR (KBr): 3417, 3075, 3038, 2934, 2909, 2834, 1620, 1590, 1513, 1465, 1346, 1300, 1229, 1173, 1079, 1031, 971, 916, 834, 802, 739. MS (ESI), m/z (% relative intensity): 288.3 (14), 316.4 (5), 454.3 (MH⁺, 100). HR MS (MH⁺) 454.18611 (calcd for C₂₆H₂₄N₅O₃ 454.18737). Anal. calcd for $C_{26}H_{23}N_5O_3{\mbox{-}1CHCl_3{\mbox{-}}}$ C, 56.61; H, 4.22; N, 12.22. Found: C, 56.61; H, 4.19; N, 12.08.

N⁶, N⁶, 8, 9-Tetrakis (4-methoxyphenyl) adenine (15aa). Yield 16%, colorless crystals from CHCl₃/heptane, R_f 0.73 (EtOAc), mp 215-218 °C. ¹H NMR (600 MHz, CDCl₃): 3.77 (s, 3H, CH₃O-4"); 3.85 (s, 6H, CH₃O-4'); 3.87 (s, 3H, CH₃O-4""); 6.71 (m, 2H, H-3",5"); 6.94 (m, 4H, H-3',5'); 7.02 (m, 2H, H-3"',5"'); 7.23 (m, 2H, H-2",6"); 7.25 (m, 2H, H-2"",6""); 7.29 (m, 4H, H-2',6'); 8.43 (s, 1H, H-2). ¹³C NMR (151 MHz, CDCl₃): 55.2 (CH₃O-4"); 55.5 (CH₃O-4'); 55.6 (CH₃O-4'''); 113.6 (CH-3'',5''); 114.3 (CH-3',5'); 115.0 (CH-3''',5'''); 121.0 (C-5); 122.0 (C-1''); 128.3 (C-1'''); 128.4 (CH-2',6'); 128.7 (CH-2"',6"'); 130.4 (CH-2",6"); 138.1 (C-1'); 148.8 (C-8); 152.6 (CH-2); 154.0 and 154.6 (C-4,6); 157.6 (C-4'); 159.7 (C-4"'); 160.5 (C-4"). IR (KBr): 2835, 1609, 1574, 1508, 1455, 1418, 1341, 1293, 1250, 1179, 1107, 1091, 1030, 838, 745, 712, 634. MS (ESI), m/z (% relative intensity): 288.3 (5), 316.4 (3), 376.3 (2), 560.3 (MH⁺, 100). HR MS (MH⁺) 560.2286 (calcd for C₃₃H₃₀N₅O₄ 560.2292). Anal. calcd for C₃₃H₂₉N₅O₄•0.1 CHCl₃: C, 69.56; H, 5.13; N, 12.25. Found: C, 69.83; H, 5.10; N, 12.29.

Single Crystal X-ray Structure Analysis. X-ray crystallographic analysis of single crystals of 7eaa (colorless, $0.14 \times 0.19 \times 0.44$ mm), 15aa (colorless, $0.18 \times 0.29 \times 0.60$ mm), and 13ad (light brown, $0.18 \times 0.31 \times 0.50$ mm) was performed with Xcalibur X-ray diffractometr with Cu_{Ka} (λ =1.54180 Å), data collected at 150K (7eaa) and 298 K (15aa, 13ad). All three structures were solved by direct methods with SIR92²⁸ and refined by full-matrix least-squares on F with CRYSTALS.²⁹ The hydrogen atoms were located in a difference map, but those attached to carbon atoms were repositioned geometrically and then refined with riding constraints, and all non-hydrogen atoms were refined anisotropically in all three cases.

Crystal Data – **7eaa.** $C_{29}H_{27}F_1N_4O_3$, orthorhombic, space group $P2_12_121$, a = 7.1004(1) Å, b = 17.4319(2) Å, c = 19.7015(2) Å, V = 2438.52(5) Å³, Z = 4, M = 498.55, 38154 reflections measured, 4993 independent reflections. Final R = 0.0317, wR = 0.0430, GoF = 1.0768 for 3587 reflections with $I > 2\sigma(I)$ and 336 parameters. CCDC 698313.

Crystal Data – **13ad.** C₂₁H₂₁N₅O₄.CHCl₃, monoclinic, space group $P_{2_1/c}$, a = 10.1515(12) Å, b = 13.1309(11) Å, c = 18.5570(17) Å, $\beta = 91.195(9)$ °, V = 2473.1(4) Å³, Z = 4, M = 526.80, 29663 reflections measured, 5219 independent reflections. Final R = 0.0488, wR = 0.0780, GoF = 1.1339 for 2510 reflections with $I > 2\sigma(I)$ and 308 parameters. CCDC 698558.

Crystal Data – **15aa.** $C_{33}H_{29}N_5O_4$, triclinic, space group $P1^-$, a = 9.8920(8) Å, b = 10.9199(7) Å, c = 15.2855(9) Å, $\alpha = 97.982(5)^\circ$, $\beta = 101.901(6)^\circ$, $\gamma = 112.999(7)^\circ$, V = 1442.4(2) Å³, Z = 2, M = 559.61, 44274 reflections measured, 6065 independent reflections. Final R = 0.0403, wR = 0.0554, GoF = 1.0621 for 4137 reflections with $I > 2\sigma(I)$ and 380 parameters. CCDC 698314.

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Supporting Information Available: Complete experimental part and characterization of all compounds, copies of all NMR spectra and cif files for **7eaa**, **13ad**, and **15aa** are provided. This material is available free of charge via the Internet at http://pubs.acs.org.

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